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## Asymmetric Reduction of Karahanaenone with Various Microorganisms

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Abstract-Both enantiomers of karahanaenol have been prepared by microbial asymmetric reduction of karahanaenone. The (S)-enantiomer with high enantiomeric excess was obtained by Glomerella cingulata, Aspergillus niger, Streptomyces aureofaciens and Bacillus subtilis respectively. In contrast, the (R)-enantiomer of karahanaenol was obtained with Fusarium solani.

Karahanaenone is a monoterpene present in hop oil. This unique and peculiar compound with its fragrance and flavour is one of the most important compounds for the flavouring industry<sup>2</sup>. Our main objective is to make a compound with a unique fragrance and flavour and which possesses a fruity and a woody odor. The reduction of karahanaenone by NaBH4 has been reported<sup>3</sup>. Chemical and biological methods have been applied to the asymmetric reduction of karahanaenone into karahanaenol however the enantioselectivity was poor. In this paper, we report a novel method for reducing karahanaenone with various microorganisms into karahanaenol with excellent eantioselectivity.

We selected the following fungi Aspergillus niger II:O 4414, Glomerella cingulata<sup>4</sup>, Fusarium solani<sup>4</sup> and Rhizoctonia solani<sup>4</sup>, and the bacterium Bacillus subtils II:O 3936, and actinomyces Streptomyces aureofaciens IFO 3187. All the microorganisms reduced karahanaenone during a period of 24hrs. The results are summarized as following in the table below

Table: Microbiological Reduction of karahanaenone

	$[\alpha]_{\mathbb{D}^{20}}$	ee	Conf	Yield
Glomerella cingulata	33	100%	S	98%
Aspergillus niger	33	100%	S	98%
Rhizoctonia solani	28	70%	S	9777
Streptomyces aureofaciens	3,3	100%	S	96%
Bacillus subtilis	33	100%	S	98%
Fusarium solani	- 28	70%	R	80%

100℃ e.e.

The absolute configuration of the secondary alcohol was determined from esters obtained by the reaction of each karahanaenol (S)-(-) and (R)-(+)- $(\alpha)$ -MTPA chlorides according to Mosher's methods<sup>5,6</sup>. Enantiomerically pure karahanaenol has not been described in the literature before.

Both enantiomers of karahanaenol have a mint like odor, The (S)-enantiomer had a notably fresher odor than the (R).

## **Experimental**

Reduction on analytical scale. Fungi case: Mycelia were respectively transplanted to the culture medium (15ml in a 50ml Petri dish) and furthermore incubated at 27°C under a static situation for 3days. After the growth of the fungus, the substrate was added directly to each medium (7.5mg/15ml) and further incubated at the same condition for 24h. Bacteria and Actinomyces case: Mycelia were then transplanted to the culture medium (100ml in a 200ml Erlenmeyer flask) and stirred for 3days at 28°C. After the growth of the fungus, the substrate was added directly to each medium (50mg/100ml) and were further incubated at the same condition for 24h. Nutrient broth for microbial transformation: A. niger. F. solani, G. cingulata: polypeptone 5g, KCl 0.5g, MgSO4·7H2O 0.01g, succharose 15g, glucose 15g, H2O 11. R. solani: glucose 50g, KNO3 10g, K2HPO4 5g, MgSO4·7H2O 2.5g, FeCl3·6H2O 0.02g, H2O 11. B. subtilis: polypeptone 5g, Yeast extract 0.2g, MgSO4·7H2O 0.1g, H2O 11. S. aureofaciens: oatmeal 5g, H2O 11. Aliquots were withdrawing periodically and monitored by GLC on a chiral column: CHROMPACK WCOAT Fused silica Cp-Cyclodextrin-B-236-M-19(50m×0.25mm, N2 flow rate 1ml/min).

Reduction on preparative scale. Karahanenone was added to the resulting mycelia of the grown cell by a process similar to the analytical scale. After 24hrs, the reaction mixture was extracted with dichloromethane with a continuous liquid-liquid extractor and concentrated under reduced pressure. The products were purified by column chromatography (silica gel, hexane-ethyl acetate ratio) to give the enantiomerically pure alcohol. Optical rotations were measured on a Japan Spectroscopic CO. LTD DIP-140 in CHCl3. The enantiomeric excesses are determined by GLC on chiral column [ retention time (min) of alcoholos: (18)-13.7, (1R)-13.4].

Determination of absolute configuration of ( ) karahanaenol. [  $\triangle \delta = \delta(-)$ MTPA- $\delta(+)$ MTPA ;(C1) proton; 4.88-4.91=-0.03, (C2) gem Methyl; 0.93-0.86=+0.07, 0.83-0.81=+0.02 (C4) proton; 5.35-5.34=+0.01, (C5) Methyl; 1.71-1.74=-0.03, (C6) proton; 2.08-2.13=-0.05, 1.95-2.01=-0.06].

## References and Notes

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